

Related topics

Peltier effect, heat pipe, thermoelectric e.m.f., Peltier coefficient, cooling capacity, heating capacity, efficiency rating, Thomson coefficient, Seebeck coefficient, Thomson equations, heat conduction, convection, forced cooling, Joule effect.

Principle

The cooling capacity) heating capacity and efficiency rating of a Peltier heat pump are determined under different operating conditions.

Equipment

Thermogenerator	04366.00	1
Flow-through heat exchanger	04366.01	1
Air cooler	04366.02	1
Heating coil with sockets	04450.00	1
Distributor	06024.00	1
Rheostat, 33 Ohm, 3.1 A	06112.02	1
Connecting plug, 2 pcs.	07278.05	1
Power supply, universal	13500.93	1
Digital multimeter	07134.00	4
Stopwatch, digital, 1/100 sec.	03071.01	1
Cold a. hot air blower, 1700 W	04030.93	1
Lab thermometer, -10+100°C	38056.00	1

Thermometer, -10+ 50°C	38033.00	2
Rubber tubing, i.d. 6 mm	39282.00	1
Universal clamp	37715.00	1
Tripod base -PASS-	02002.55	1
Support rod -PASS, $l = 250 \text{ mm}$	02025.55	1
Right angle clamp -PASS-	02040.55	1
Connecting cord, $l = 250$ mm, red	07360.01	3
Connecting cord, $l = 500$ mm, red	07361.01	3
Connecting cord, $l = 500$ mm, blue	07361.04	2
Connecting cord, $l = 750$ mm, blue	07362.04	2
Connecting cord, $l = 750$ mm, red	07362.01	1
Heat conductive paste, 50 g	03747.00	1

Tasks

- 1. To determine the cooling capacity P_c the pump as a function of the current and to calculate the efficiency rating η_c at maximum output.
- 2. To determine the heating capacity P_w of the pump and its efficiency rating η_w at constant current and constant temperature on the cold side.
- 3. To determine $P_{\rm w}$, $\eta_{\rm w}$ and $P_{\rm c}$, $\eta_{\rm c}$ from the relationship between temperature and time on the hot and cold sides.
- 4. To investigate the temperature behaviour when the pump is used for cooling, with the hot side air-cooled.

Fig. 1: Experimental set-up for measuring cooling capacity.

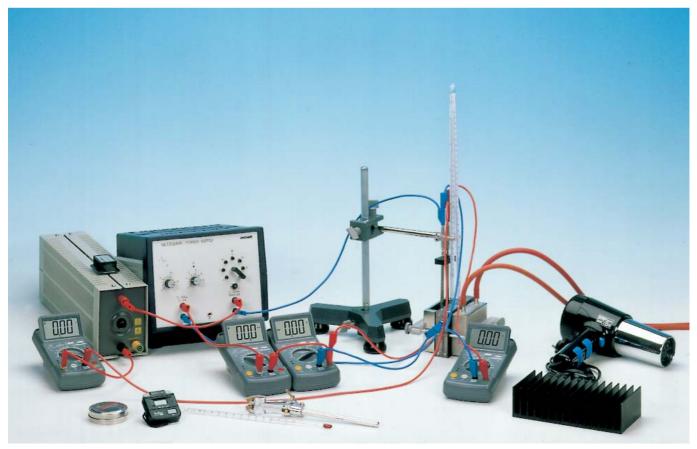
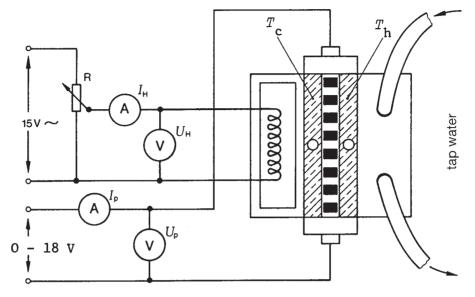




Fig. 2: Set-up for determining cooling capacity.



Set.up and procedure

- 1. Fit a water bath on the cold side and a heat exchanger through which tap water flows on the hot side. A heating coil (resistance approx. 3 ohms), operated on AC, dips into the water-filled bath. For each current value I_p set the heating capacity $P_H = U_H \cdot I_H$ with the rheostat R so that the temperature difference between the hot and the cold side is approximately zero. The power supplied then exactly corresponds to the cooling capacity P_c . Measure the heater current I_H and voltage U_H , the operating current I_p and the cold side T_c .
- 2. Remove the heating coil as it is no longer required. Reverse the operating current so that the water in the bath now heats up. Measure the rise in the temperature of water $T_{\rm w}$ at constant current $I_{\rm p}$. Measure also $I_{\rm p}$, $U_{\rm p}$ and $T_{\rm c}$. Calculate the heat capacities of a copper block $C_{\rm Cu}$, of the water $C_{\rm w}$ and of the brass bath $C_{\rm Br}$ from their dimensions or by weighing.
- 3. Fit water baths to both sides of the heat pump and fill them with water of the same temperature. With the current I_p constant) measure the changes in the temperature of the two water baths) i.e. $T_h = f(t)$, $T_c = f(t)$, I_p and U_p .
- For this fourth experiment we have a water bath on the cold side, an air cooler on the hot. Measure the temperature of the cold side as a function of time, with the cooler a) in static atmospheric air, and
 - b) force-cooled with a blower.

Theory and evaluation

When an electric current flows through a circuit composed of two different conductors, heat will be liberated at one junction and absorbed at the other) depending on the direction in which the current is flowing (Peltier effect). The quantity of heat Q liberated per unit time is proportional to the current I:

$$\frac{Q}{t} = P_{p} = \pi \cdot I = \alpha \cdot T \cdot I$$

where π is the Peltier coefficient, α the Seebeck coefficient and *T* the absolute temperature.

If an electric current I flows in a homogeneous conductor in the direction of a temperature gradient

$$\frac{\mathrm{d}T}{\mathrm{d}x}$$

heat will be absorbed or given out, depending on the material (Thomson effect):

$$P_{\mathsf{T}} = \tau \cdot I \cdot \frac{\mathsf{d}T}{\mathsf{d}x}$$

where τ is the Thomson coefficient.

The direction in which the heat flows depends on the sign of the Thomson coefficient, the direction in which the current flows and the direction of the temperature gradient.

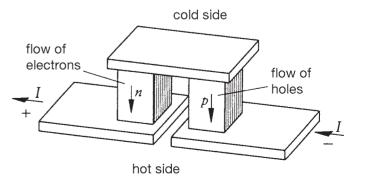
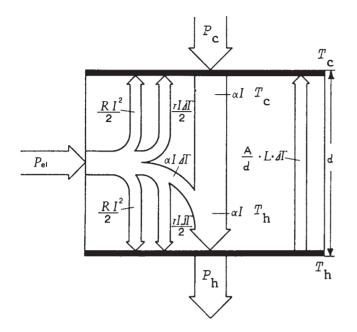


Fig. 3: Construction of a Peltier semi-conductor element. In practice several elements are generally connected in series (electrically) and in parallel (thermally).



Fig. 4: Power balance flow chart in a Peltier component. (The example illustrated is for the case where $P_{\rm T}$ > 0).



If an electric current I flows in an isothermal conductor of resistance R, we have the Joule effect:

$$P_{.1} = R \cdot I^2$$

Because of heat conduction, heat also flows from the hot side (temperature $T_{\rm h}$) to the cold side (temperature $T_{\rm c}$):

$$P_{\rm L} = L \frac{A}{d} (T_{\rm h} - T_{\rm c})$$

where L is the conductivity, A the cross-sectional area and d the thickness of the Peltier component.

Writing $\Delta T = T_h - T_c$, we obtain for the heat capacity of the pump on the cold side (the cooling capacity):

$$-P_{\rm c} = \alpha T_{\rm c} I \pm \frac{\tau I \Delta T}{2d} - \frac{1}{2} I^2 R - \frac{L \cdot A \cdot \Delta T}{d}$$

and, for the heat capacity of the pump on the hot side (the heating capacity):

$$+P_{\rm h} = \alpha T_{\rm h} I \pm \frac{\tau I \Delta T}{2d} + \frac{1}{2} I^2 R - \frac{L \cdot A \cdot \Delta T}{d}.$$

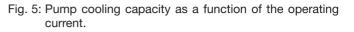
The electric power supplied is

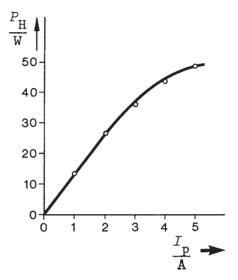
$$+P_{\rm el.} = \alpha I \Delta T + RI^2 + \frac{\tau I \Delta T}{2d} = U_{\rm p} \cdot I_{\rm p}$$

1. The pump cooling capacity $P_{\rm c}$ was found to be 49 W when $I_{\rm p}$ = 5 A and $P_{\rm H}$ = $P_{\rm c}$

The efficiency rating

 $\eta_{c} = \frac{P_{c}}{P_{el.}}$





becomes, for the measured values $I_{\rm p} = 5.0$ A and $U_{\rm p} = 14.2$ V, $\eta_{\rm c} = 0.69 (\vartheta_{\rm h} = \vartheta_{\rm c} = 20$ °C)

2. From the slope of the curve in Fig. 6 (where the curve starts off as a straight line) we can calculate the pump heating capacity

$$P_{\rm h} = \frac{C_{\rm tot.} \cdot \Delta T_{\rm h}}{\Delta t}$$

and the corresponding efficiency rating

$$\eta_{\rm h} = \frac{P_{\rm h}}{P_{\rm el.}} \; ,$$

where $P_{\rm el.} = I_{\rm p} \cdot \overline{U}_{\rm p}$

as follows:

$$m_{\rm w} = 0.194 \text{ kg}, \qquad c_{\rm w} = 4182 \frac{J}{\text{ kg K}}$$

$$m_{\rm Br} = 0.0983 \text{ kg}, \qquad c_{\rm Br} = 381 \frac{J}{\text{ kg K}}$$

$$m_{\rm Cu} = 0.712 \text{ kg}, \qquad c_{\rm Cu} = 383 \frac{J}{\text{ kg K}}$$

$$c_{\rm tot.} = m_{\rm w} \cdot c_{\rm w} \cdot m_{\rm Br} \cdot c_{\rm Br} + m_{\rm Cu} \cdot c_{\rm Cu} = 1121 \frac{J}{\text{ kg K}}$$

where $m_{\rm W}$ is the mass of the water, $c_{\rm W}$ the specific heat capacity of the water, $m_{\rm Cu}$ the mass of a copper block, $c_{\rm Cu}$ the specific heat capacity of copper, $m_{\rm Br}$ the mass of the brass bath, $c_{\rm Br}$ the specific heat capacity of brass, $I_{\rm p}$ the pump current, and $\overline{U}_{\rm p}$ the mean pump voltage.

With the slope

$$\frac{\Delta T_{\mathsf{h}}}{\Delta t}$$
 = 6.7 × 10⁻² K/s

we obtain a value $P_{\rm h}$ of 75 W.

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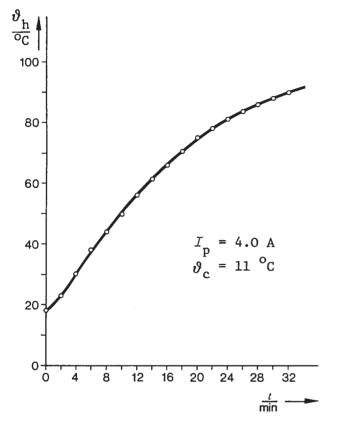


Fig. 6: Temperature of the hot side as a function of time.

With values for $I_{\rm p}$ of 4.0 A and $\overline{U}_{\rm p}$ of 12.5 V (average value) we obtain an efficiency rating

 $\eta_{c} = 1.5$

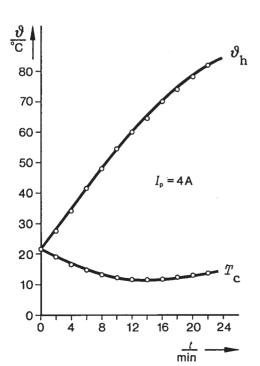
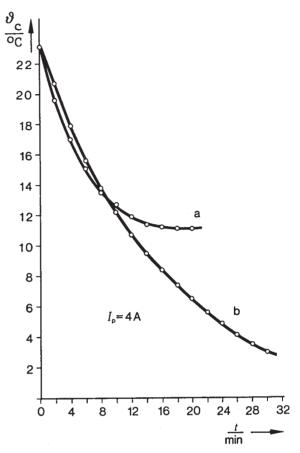


Fig. 7: Water temperature as a function of time.

- Fig. 8: Water temperature when the hot side is cooled with an air cooler
 - a) cooling by convection
 - b) forced cooling.



3. $P_{\rm h}$ and $P_{\rm c}$, and $\eta_{\rm h}$ and $\eta_{\rm c}$, can be calculated from the slopes of the curves $\vartheta_{\rm h} = f(t)$ and $\vartheta_{\rm c} = f(t)$ and the relevant heat capacities.

With $\frac{\Delta \vartheta_{\rm h}}{\Delta t}$ = 0.056 K/s (start of curve) and

$$\frac{\Delta \vartheta_{\rm c}}{\Delta t}$$
 c = – 0.023 K/s and with $C_{\rm tot.}$ = 1121 J/K, we obtain:

$$P_{\rm h} = 63 \text{ W}; \qquad P_{\rm c} = 26 \text{ W}.$$

In the range considered, the voltage \overline{U}_{p} (average value) was 12.4 V, so that we obtain the efficiency ratings $\eta_{h} = 1.3$ and $\eta_{c} = 0.52$. ($I = 4 \text{ A}, T = 22^{\circ}\text{C}$).

4. Fig. 8 shows the course of temperature in the water bath on the cold side when the hot side was cooled with the air cooler. The temperature ϑ_h of the hot side was approx. 72°C after 20 minutes (no blower). The maximum temperature difference $\vartheta_h - \vartheta_c = 60$ K is thus attained and the pump output of the Peltier component is zero. When the blower was used, T_h remained constant at approx. 45°C after 20 minutes.